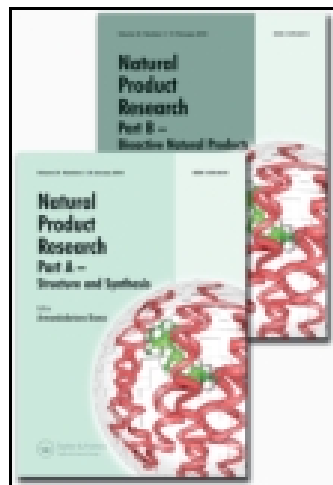


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## NEW ANTIBACTERIAL STEROIDAL ALKALOIDS FROM *SARCOCOCCA BREVIFOLIA*

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**Abstract:** Three new steroidal alkaloids, epipachysamine-*E*-5-ene-4-one (1), *N*<sub>B</sub>-demethylepipachysamine-*E*-5-ene-4-one (2) and *iso-N*-formylchonemorphine (3) have been isolated from *Sarcococca brevifolia*. Structures of these compounds were determined by spectroscopic studies. Compounds 1 and 3 exhibited strong antibacterial activity against *Bacillus cereus*, *Escherichia coli*, *Staphylococcus aureus*, *Klebsiella pneumoniae*, *Proteus mirabilis*, *Corynebacterium diphtheriae* and *Pseudomonas aeruginosa*.

**Key Words:** Steroidal alkaloids; *Sarcococca brevifolia*; pregnane; epipachysamine-*E*-5-ene-4-one; *N*<sub>B</sub>-demethylepipachysamine-*E*-5-ene-4-one; *iso-N*-formylchonemorphine; antibacterial.

### INTRODUCTION

The plants of family Buxaceae are well known for producing steroidal alkaloids and a number of new steroidal alkaloids have been reported.<sup>1-12</sup> It comprises several genera including *Pachysandra*, *Sarcococca*, *Simondsia* and *Buxus* which are rich sources of steroidal alkaloids.<sup>13</sup> These plants have been extensively used for the treatment of malaria, rheumatism and skin diseases.<sup>1-3, 14</sup>

There are two *Sarcococca* species, *S. brevifolia* and *S. ceylanica* occurring in Sri Lanka and both are endemic.<sup>15</sup> *S. brevifolia* Muel.Arg. is an evergreen shrub, growing in the central of the country. No phytochemical work has reported on methanolic extract of this plant so far. Present studies on the ethanolic extract of the plant has yielded three new steroidal alkaloids epipachysamine-*E*-5-ene-4-one (1), *N*<sub>B</sub>-demethylepipachysamine-*E*-5-ene-4-one (2) and *iso-N*-formylchonemorphine (3). Structures of these compounds were established through detailed spectroscopy such as 1D and 2D NMR experiments.<sup>16</sup>

## RESULTS AND DISCUSSION

Compound (1) was isolated as an amorphous solid. The HREIMS of (1) exhibited the  $M^+$  at  $m/z$  440.3304 analyzing for  $C_{28}H_{44}N_2O_2$  (calc. 440.3403), hence the compound (1) possessed eight degrees of unsaturation, four of which were accounted for by the tetracyclic pregnane-type steroidal skeleton, two were due to carbonyl groups and the remaining two were due to olefinic bonds. Compound (1) showed IR absorptions at 3651 (NH), 1660 (amide), 1630 and 1490 (CO conj.sec.amide)  $cm^{-1}$ . The  $^1H$ -NMR spectrum of (1) showed two one-proton multiplets at  $\delta$  5.67 and 7.67 due to the C-2' and C-6 olefinic protons, respectively. A six-proton singlet at  $\delta$  2.16 was ascribed to  $N_b(Me)_2$  substituted at C-20, a three-proton doublet at  $\delta$  0.98 ( $J = 6.5$  Hz) was due to the C-21 secondary methyl protons and a one-proton multiplet at  $\delta$  1.90 was due to the C-20 methine proton. Another one-proton multiplet at  $\delta$  3.50 was assigned to the C-3 methine proton while two (three-proton) methyl singlets at  $\delta$  0.65 and at  $\delta$  0.85 were due to angular C-18 and C-19 methyl groups respectively.

The  $^{13}C$ -NMR (broad-band decoupled and DEPT spectra) of (1) revealed 7 signals for  $CH_2$ , 8 signals for  $CH$ , 7 signals for  $CH_3$  and 6 signals for quaternary carbons. The signal at  $\delta$  196.2 was assigned to  $\alpha$ ,  $\beta$ -unsaturated carbonyl carbon (C-4), while that at  $\delta$  165.5 was assigned to the amidic carbonyl carbon (C-1'). Other important signals were at  $\delta$  131.6 (C-5), 125.6 (C-6), 153.0 (C-3'), and 118.7 (C-2'). The complete  $^1H$ - and  $^{13}C$ -NMR chemical shift assignments are summarized in Table-1. The HMBC and COSY-45° techniques also confirmed these assignments.

The UV, IR and  $^1H$ -NMR spectra of compound (2) ( $C_{27}H_{42}N_2O_2$ ) were similar to those of (1) except that the  $^1H$ -NMR spectrum of (2) exhibited only a 3H singlet at  $\delta$  2.16 due to the  $N_b$ -methyl group. The spectroscopic studies therefore showed that (2) is  $N_b$ -demethyl derivative of (1).

Compound (3) was also isolated as an amorphous solid. EIMS and FDMS showed the molecular ion at  $m/z$  374. HREIMS exhibited the  $M^+$  peak at  $m/z$  374.3297 analyzing for  $C_{24}H_{42}N_2O$  (calc. 374.3296), hence it possessed five degrees of unsaturation, four of which were due to the tetracyclic structure of the pregnane-type skeleton and one was due to the presence of a carbonyl group present in it. Compound (3) showed the base peak at  $m/z$  72 representing fragment  $C_4H_6NO$  which arose by the cleavage of C-17/C-20 bond.

The  $^1H$ -NMR spectrum of (3) showed a one-proton singlet at  $\delta$  8.07 was due to the aldehydic proton, another one-proton signal at  $\delta$  5.30 was due to the NH proton. A three-proton doublet at  $\delta$

1.17 ( $J = 6.5$  Hz) was ascribed to the C-21 methyl and a one-proton multiplet at  $\delta$  4.10 was assigned to the C-20 methine proton. A one-proton multiplet at  $\delta$  3.50 was assigned to the C-3 methine proton and a six-proton singlet at  $\delta$  2.50 was due to  $N_a(Me)_2$  group. The HMBC and COSY-45° connectivities confirmed these assignments. The  $^{13}C$ -NMR spectra showed 9 signals for  $CH_2$ , 7 signals for CH, 5 signals for  $CH_3$  and 3 signals for quaternary carbon atoms. The signal at  $\delta$  159.0 was assigned to amidic carbonyl carbon atom. Compound (3) is the isomer of  $N_a$ -formyl chonemorphone which has  $N$ -formyl substituent at C-3<sup>17</sup>. The complete  $^1H$ - and  $^{13}C$ -NMR chemical shift assignments have been summarized in Table-1.

Compounds (1) and (3) exhibited strong activity against a number of pathogenic bacteria. Antibacterial activity of (1) was against *Bacillus cereus*, *Klebsiella pneumoniae* and *Staphylococcus aureus*. Similarly compound (3) has shown strong activity against *Bacillus cereus*, *Pseudomonas aeruginosa*, *Staphylococcus aureus*. Due to their strong antibacterial activities comparable to that of standard antibiotics such as amoxicillin and ampicillin, cephalexin etc. they can be likely candidates for *in vivo* evaluation. MIC values of both of these compounds are given in Table-2.

#### GENERAL EXPERIMENTAL

IR spectra were recorded on JASCO 302-A spectrophotometer. UV spectra were recorded on Hitachi U-3200 spectrophotometer. EI MS (70 eV), FD, HREI MS were measured on JMS-HX 100 and JMS-DA 500 mass spectrophotometers.  $^1H$ -NMR and  $^{13}C$ -NMR spectra were recorded on Bruker NMR spectrometers operating at 500 and 125 MHz respectively at room temperatures. The chemical shift values are reported in ppm ( $\delta$ ) units. Standard pulse sequences were used for COSY-45°, HOHAHA, DEPT, HMQC and HMBC experiments.

Merck silica gel GF 254 (0.25 mm) and silica PF 254 Merck (1 mm) were used for thin layer chromatographic separations. The eluent for compounds (1-3) was pure  $CHCl_3$ . Spots and bands were detected under UV radiations (254 and 366 nm). Spray reagent for TLC was Dragendorff's reagent.

#### EXTRACTION AND ISOLATION

The dried whole plant of *Sarcococca brevifolia* was extracted with methanol. Evaporation of the methanol extract gave a brown solid (275 g) which was dissolved in 2N HCl and the acidic extract was washed with  $CH_2Cl_2$ . Basification with 20%  $NH_4OH$  followed by extraction with  $CH_2Cl_2$

gave the non-quaternary alkaloids (2 g) as a brown solid. Separation of the latter over silica gel (Merck Art 7734; eluent: hexane-ethyl acetate-methanol) followed by preparative thin-layer chromatography (eluent  $\text{CHCl}_3$ , chamber saturated with  $\text{NH}_3$  vapour) gave three new steroidal alkaloids epipachysamine-*E*-5-ene-4-one (1), *N<sub>b</sub>*-demethylepipachysamine-*E*-5-ene-4-one (2) and *iso-N*-formylchonemorphone (3).

## BACTERICIDAL BIOASSAY

Bactericidal activity was determined by agar well diffusion method. This test was performed by spreading 18-24 hour old pathogenic bacterial cultures containing approximately  $10^4$ - $10^6$  colony forming units (CFU/mL) on the surface of nutrient agar (Bio M Laboratories, USA BMO 13-62-N) plates. Wells were dugged in the media with the help of a sterile metallic borer. Test samples of different concentrations prepared in dimethylsulfoxide (DMSO Merck) are added in their respective wells. Pure DMSO was used as a control. Other wells are supplemented with reference compounds *i.e.* amoxicillin. $3\text{H}_2\text{O}$ , ampicillin. $3\text{H}_2\text{O}$  and cephalixin- $\text{Na}^+$ , serving as positive controls.<sup>18</sup>

## EXPERIMENTAL

**Epipachysamine-*E*-5-ene-4-one (1)** (30 mg,  $1.5 \times 10^{-3}$  %),  $\text{C}_{28}\text{H}_{44}\text{N}_2\text{O}_2$ ; amorphous solid *m.p.* 150-151°C;  $[\alpha]_{\text{D}}^{25} = 6^\circ$  ( $c = 0.2$ ,  $\text{CHCl}_3$ ),  $R_f = 0.51$ ; UV (MeOH)  $\lambda_{\text{max}}$ ; 194 (log  $\epsilon = 4.11$ ); IR ( $\text{CHCl}_3$ )  $\nu_{\text{max}}$ ; 3651, 3325, 1660, 1630, 1490  $\text{cm}^{-1}$ , HREI MS  $m/z$ ; 440.3403 (calc.  $\text{C}_{28}\text{H}_{44}\text{N}_2\text{O}_2$ ); EI MS:  $m/z$  (rel. int., %) 440 (3), 425 (5), 84 (47), 72 (100),  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR see Table 1.

***N<sub>b</sub>*-Demethylepipachysamine-*E*-5-ene-4-one (2)** (10 mg,  $5 \times 10^{-4}$ );  $\text{C}_{27}\text{H}_{42}\text{N}_2\text{O}_2$ ; amorphous solid *m.p.* 145-147°C,  $[\alpha]_{\text{D}}^{25} = -8^\circ$  ( $c = 0.29$ ,  $\text{CHCl}_3$ );  $R_f = 0.9$ ; UV (MeOH)  $\lambda_{\text{max}}$  194 (log  $\epsilon = 4.11$ ); IR ( $\text{CHCl}_3$ )  $\nu_{\text{max}}$ ; 3651, 3325, 1660, 1630, 1490  $\text{cm}^{-1}$ , HREI MS;  $m/z$  426.3246 (calc.  $\text{C}_{27}\text{H}_{42}\text{N}_2\text{O}_2$ ); EI MS:  $m/z$  (rel. int., %) 426 (7), 369 (15), 83 (100);  $^1\text{H}$ -NMR ( $\text{CDCl}_3$ , 500 MHz),  $\delta$  3.50 (1H, m, H-3), 7.67 (1H, m, H-6), 0.98 (1H, d,  $J = 6.5$  Hz, H-21), 2.16 (3H, s,  $\text{N}_b\text{CH}_3$ ), 2.07 (1H, br.d.,  $\text{N}_b\text{H}$ ), 5.30 (1H, br.d.,  $\text{N}_a\text{H}$ ), 5.67 (1H, s, H-2'), 1.85 (6H, s, 2Me at C-3').

***Iso-N*-Formylchonemorphone (3)** (50 mg,  $2.5 \times 10^{-3}$  %);  $\text{C}_{24}\text{H}_{42}\text{N}_2\text{O}$ ; amorphous solid *m.p.* 220-222°C,  $[\alpha]_{\text{D}}^{25} = -14^\circ$  ( $c = 0.2$ ,  $\text{CHCl}_3$ ),  $R_f = 0.36$ ; UV (MeOH)  $\lambda_{\text{max}}$  192 (log  $\epsilon = 3.80$ ); IR ( $\text{CHCl}_3$ )  $\nu_{\text{max}}$ ; 3400, 1678, 1500  $\text{cm}^{-1}$ ; HREI MS;  $m/z$  374.3297 (calc.  $\text{C}_{24}\text{H}_{42}\text{N}_2\text{O}$ ); EI MS:  $m/z$  (rel. int., %) 374 (18), 359(2), 317(2), 302(2), 147(1), 110 (49), 84(100),  $^1\text{H}$ - and

$^{13}\text{C}$ -NMR  $\delta$  see Table 1.

**Table-1:**  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectra for 1 and 3 ( $\text{CDCl}_3$ , 500 MHz).

1			3	
Carbon	$^{13}\text{C}$ -NMR	$^1\text{H}$ -NMR	$^{13}\text{C}$ -NMR	$^1\text{H}$ -NMR
1	20.2 ( $\text{CH}_2$ )	0.97/2.15(m)	36.9 ( $\text{CH}_2$ )	1.06/1.87(m)
2	20.5 ( $\text{CH}_2$ )	1.32/2.05(m)	22.1 ( $\text{CH}_2$ )	2.02/1.65(m)
3	54.6 ( $\text{CH}$ )	3.50 (m)	65.4 ( $\text{CH}$ )	3.50 (m)
4	196.2 (C)	-	39.4 ( $\text{CH}_2$ )	1.14/2.0(m)
5	153.0 (C)	-	53.9 ( $\text{CH}$ )	2.02 (m)
6	125.6 ( $\text{CH}$ )	7.67 (m)	28.4 ( $\text{CH}_2$ )	1.32/1.75 (m)
7	39.0 ( $\text{CH}_2$ )	2.25/2.23 (m)	23.9 ( $\text{CH}_2$ )	1.09/1.59 (m)
8	50.1 ( $\text{CH}$ )	2.55 (m)	35.2 ( $\text{CH}$ )	1.35 (m)
9	31.8 ( $\text{CH}$ )	2.05 (m)	56.5 ( $\text{CH}$ )	1.06 (m)
10	34.8 (C)	-	35.9 (C)	-
11	24.0 ( $\text{CH}_2$ )	1.60/1.76 (m)	20.0 ( $\text{CH}_2$ )	1.15/1.20 (m)
12	39.4 ( $\text{CH}_2$ )	1.15/2.35 (m)	39.2 ( $\text{CH}_2$ )	1.12/1.90 (m)
13	39.9 (C)	-	42.1 (C)	-
14	34.6 ( $\text{CH}$ )	1.55 (m)	45.4 ( $\text{CH}$ )	1.15 (m)
15	30.5 ( $\text{CH}_2$ )	1.80/0.85 (m)	31.7 ( $\text{CH}_2$ )	1.68/0.88 (m)
16	27.8 ( $\text{CH}_2$ )	1.60/1.90 (m)	26.8 ( $\text{CH}_2$ )	1.20/1.15 (m)
17	54.8( $\text{CH}$ )	1.30 (m)	56.4 ( $\text{CH}$ )	1.30 (m)
18	12.3 ( $\text{CH}_3$ )	0.65 (s)	12.3 ( $\text{CH}_3$ )	0.65 (s)
19	13.3 ( $\text{CH}_3$ )	0.85 (s)	12.3 ( $\text{CH}_3$ )	0.85 (s)
20	56.0 ( $\text{CH}$ )	1.90 (m)	46.7 ( $\text{CH}$ )	4.10 (m)
21	12.1 ( $\text{CH}_3$ )	0.98 (d)	21.7 ( $\text{CH}_3$ )	1.17 (d, $J = 6.5$ Hz)
$\text{N}(\text{CH}_3)_2$	40.0 ( $\text{CH}_3$ )	2.16 (s)	39.4 ( $\text{CH}_3$ )	2.50 (s)
NH	-	-	-	5.30 (br.d)
N-CHO	-	-	-	8.07 (s)
1'	165.5 (C)	-	159.0 (C)	-
2'	118.7 ( $\text{CH}$ )	5.67 (m)		
3'	131.6 (C)	-		
4'	18.0 ( $\text{CH}_3$ )	1.86 (s, 6H)		
5'	25.0 ( $\text{CH}_3$ )	-		

**Table-2:** Antibacterial Activity of Compounds **1** and **3** (Minimal inhibitory concentration in mg/ml).

Bacteria	MIC		Standard Drugs	MIC
	<b>1</b> (mg/ml)	<b>3</b> (mg/ml)		
<i>B. cereus</i>	0.0625	0.1250	Amoxicillin. 3H <sub>2</sub> O	0.2500
			Ampicillin. 3H <sub>2</sub> O	0.2500
			Cephalexin-Na <sup>+</sup>	0.0078
<i>C. diphtheriae</i>	0.2500	0.2500	Amoxicillin. 3H <sub>2</sub> O	0.2500
			Ampicillin. 3H <sub>2</sub> O	0.1250
			Cephalexin-Na <sup>+</sup>	0.1250
<i>E. coli</i>	0.2500	0.2500	Amoxicillin. 3H <sub>2</sub> O	0.2500
			Ampicillin. 3H <sub>2</sub> O	1.0000
			Cephalexin-Na <sup>+</sup>	0.5000
<i>K. pneumoniae</i>	0.2500	0.2500	Amoxicillin. 3H <sub>2</sub> O	0.2500
			Ampicillin. 3H <sub>2</sub> O	0.2500
			Cephalexin-Na <sup>+</sup>	0.2500
<i>P. mirabilis</i>	0.2500	0.2500	Amoxicillin. 3H <sub>2</sub> O	0.00781
			Ampicillin. 3H <sub>2</sub> O	0.00781
			Cephalexin-Na <sup>+</sup>	-
<i>P. aeruginosa</i>	0.1250	0.1250	Amoxicillin. 3H <sub>2</sub> O	0.1250
			Ampicillin. 3H <sub>2</sub> O	0.0625
			Cephalexin-Na <sup>+</sup>	-
<i>S. aureus</i>	0.03125	0.0312	Amoxicillin. 3H <sub>2</sub> O	0.2500
			Ampicillin. 3H <sub>2</sub> O	0.1250
			Cephalexin-Na <sup>+</sup>	-

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